

Organic Peroxide For Polymer & Rubber Elastomers & Thermoplastics



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DO SENDER CHEM

We are a professional enterprise that researches, develops, produces, and sells organic peroxide initiators, curing agents, catalysts, and vulcanizing agents. The company has a sound product quality assurance system, advanced production technology, and complete analytical and experimental instruments. With advanced technology, efficient management, and trustworthy products, it sells well throughout the country and is exported worldwide.

For years, we have been providing the Organic Peroxides to the global,. High quality products cannot be separated from a high-quality safety management system. Our production facilities are designed by a team of top domestic chemical design institutes, which keep up with the international first-class level in terms of safety properties and process flow. At the same time, our R&D team is constantly communicating with top universities and research institutes at home and abroad.





The process technology of organic peroxide series products has been operating safely for many years. This production technology is mature, reliable, advanced, with high product quality and low production pollution. It is currently widely used in relevant domestic enterprises. This project has a high level of automation and machinery, and the equipment is well sealed. It adopts a DCS automation control system for centralized control.

For hazardous chemicals that constitute major hazard sources and key monitored hazardous chemical processes, an SIS safety instrument system is also installed to centrally detect, display, interlock, control, and alarm during the production process of the device. The construction unit has many years of production experience in this product, ensuring the safety and reliability of the entire process.



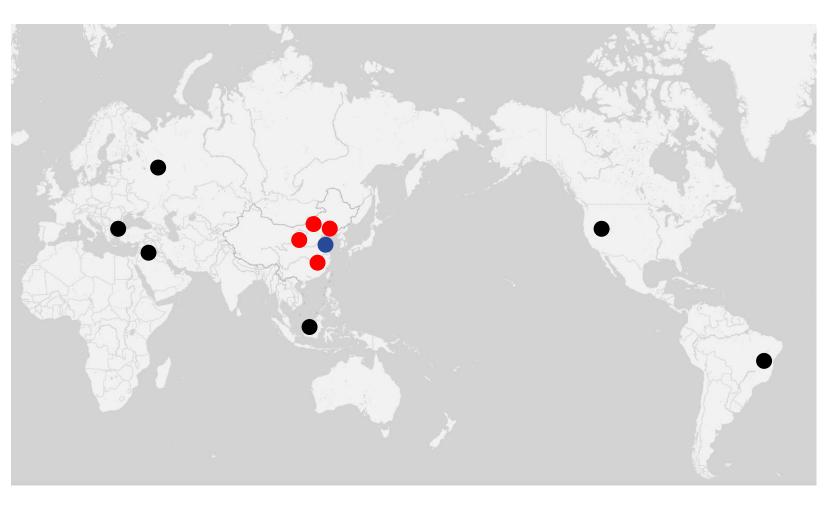
We Create Valuable Passions

Together, we bring essential solutions for a sustainable future

We are a global, specialty chemicals leader. Markets and consumers worldwide rely on our essential solutions to manufacture everyday products, such as cleaning goods, paints and coatings and building products. Furthermore, the dedication of approximately 500 employees with a shared commitment to our customers, business growth, safety, sustainability and innovation safety and high-quality reliability of our has resulted in a consistently strong financial performance. Within our Polymer Specialties business, we produce everyday essentials for the global polymer, recycling and polymer processing industries. We are among the world's leading producers of organic peroxides, metal alkyls and organometallic specialties, which are essential ingredients for the thermoplastic, composite and rubber industries.

Global Service Network

Although our manufacturing facilities are based on the mainland of China, Our strong resource integration capability enables us to provide more reliable, sustainable, and reliable services in the current international situation. For the supply chain, every node is crucial, and it is also the key to ensuring the products. All of our factories have passed ISO9001 certifications to ensure the highest product quality and strict compliance with environmental regulations. In addition, we continue to increase investment in manufacturing technology, high-quality standards, safety, innovation, effective technical support, and reliable supply chains.



Headquarter

Branch

Manufacturing Facility

Lead to a Sustainable Future

We partner with our customers, suppliers and employees to deliver innovative solutions, drive progress and create a safe and sustainable today and tomorrow for everyone.

Our Commitment to a Sustainable Future, is based on following pillars:









We are one of the companies in China that produces and operates the largest variety of organic peroxide products, with the largest trade volume and the best product quality. The company has the production capacity for 30 types of organic peroxide products, while other Class 5.2 products can be sold and exported. A strong supply chain and high-quality service provide the possibility for this. Our current factory facility has an annual production capacity of approximately 10,000 tons of organic peroxides, which is superior to other Chinese suppliers







Productivity 1000T/A

Factory Cover Area
2000m²

Investment
20M

Product Quantity
50+





Range of Crosslinking Peroxides

Do Sender Chem's range of organic peroxides for the crosslinking of elastomers and thermoplastics is very extensive. Companies all over the world depend on our **Perodox®** organic peroxide brands. Why? Because they are an important ingredient in the production of products, ranging from hi-tech automotive parts such as hoses and belts to shoe soles and power distribution cables.



Examples include:

• Perodox® 101

PEX pipes, polymer modification, technical rubber goods

• Perodox® B

Polymerization of acrylate and methacrylate Perodox ® 14

wire 8 cable, technical rubber goods, footwear

Perodox®K

Polymer production and Poly(meth)acrylics

Perodox® L

initiator used for curing unsaturated polyester, vinyl ester and acrylic thermoset

• Perodox® LUNA

initiator for the suspension and mass polymerization of vinyl chloride



Much of our success is due to our philosophy of creating close partnerships with our customers. What do you want to achieve? From optimizing applications, improving efficiencies, resolving difficulties or even developing new crosslinking peroxides, we're happy to meet with you to discuss your requirements.

This product guide provides an overview of our main, commercially available crosslinking peroxides. We invite you to visit us at www.perodox.com for complete product listings.

Uses of Crosslinking Peroxides

| Peroxides are used to crosslink |
|--|
| NR: Natural rubber |
| IR: Polyisoprene |
| BR: Polybutadiene |
| CR: Polychloroprene |
| SBR: Styrene butadiene rubber |
| NBR: Nitrile rubber |
| HNBR: Hydrogenated nitrile rubber |
| Q: Silicone |
| AU/EU: Polyurethane |
| EPM: Ethylene propylene copolymer |
| EPDM: Ethylene propylene terpolymer |
| POE: Polyolefin elastomer |
| T: Polysulfide |
| PE: Polyethylene |
| CM: Chlorinated polyethylene |
| CSM: Chlorosulfonated polyethylene |
| EVA: Ethylene vinylacetate copolymer |
| ABS: Acrylonitrile butadiene styrene copolymer |
| AEM: Ethylene acrylic |
| EBA: Ethylene butylacrylate copolymer |
| FKM: Fluoro elastomers |

| and blends of |
|-----------------|
| NBR/EPDM |
| SBR/EPDM |
| PP/EPDM (TPV's) |
| PE/EVA |
| NBR/EVA |
| POE/EP(D)M |

Organic peroxides also find growing use in Polymer modification (CR-PP) Recycling Grafting processes (silane, maleic anhydride) Dynamic vulcanization (TPV production)

Peroxides find limited use or cannot be used to crosslink

| ACM: Polyacrylate |
|--------------------------------|
| IIR: Butyl rubber |
| CIIR: Chlorobutyl rubber |
| CO: Epichlorohydrin |
| ECO: Epichlorohydrin copolymer |
| PP: Polypropylene |
| PB: Polybutene-1 |
| PIB: Polyisobutene |
| PVC: Polyvinylchloride |
| |



Our Peroxide Products

| Common | | AS Molecular formula | Storage Data | | | | | Class/ | | | |
|-------------------|-----------|---|--|---|--------------------------------|---------------|-------------------------------------|---------------------------------|--------------|--------|-------|
| Name | CAS | | SADT /°C | Emergency temperature (T _e) | Control temperature (Tc) | Ts Max /°C | Ts Min /°C | Description | Divison | UN No. | PG |
| Perodox B | 110-05-4 | CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 | 80°C | | | 40°C | -30°C to prevent crystallization | Organic peroxide type E; liquid | Division 5.2 | 3107 | |
| Perodox K90 | 80-15-9 | CH ₃ | 75°C and 70°C for IBCs | | | | -30°C to prevent crystallization | Organic peroxide type F; liquid | Division 5.2 | 3109 | |
| Perodox K80 | 80-15-9 | CH ₃ | 75°C for small cans, 70°C for IBC's and 65°C for bulk tanks. | | | 40°C | -30° | Organic peroxide type F; liquid | Division 5.2 | 3109 | |
| Perodox L75 | 94-36-0 | 0 0 | 80°C | 75°C | | 40°C | | Organic peroxide type C; solid | Division 5.2 | 3104 | PG II |
| Perodox L50-PS | 94-36-0 | | 70°C | | | 30°C | | Organic peroxide type E; solid | Division 5.2 | 3108 | |
| Perodox C | 614-45-9 | O CH ₃ C-O-O-C-CH ₃ CH ₃ | 60°C | | | 25°C | 10°C | Organic peroxide type C; liquid | Division 5.2 | 3103 | |
| Perodox 21 | 3006-82-4 | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 35°C | 25°C | | 10°C | -30°C to prevent crystallization | Organic peroxide type C; liquid | Division 5.2 | 3113 | PG II |



Our Peroxide Products

| Common | | | Storage Data | | | | | Class/ | | | |
|----------------------|------------|---|--------------|---|--------------------------------|---------------|-----------|--------------------------------|--------------|--------|----|
| Name | CAS | Molecular formula | SADT /°C | Emergency temperature (T _e) | Control temperature (Tc) | Ts Max /°C | Ts Min /℃ | Description | Divison | UN No. | PG |
| Perodox 14 96-PD | 2212-81-9 | | 80°C | | | 30°C | | Organic peroxide type D; solid | Division 5.2 | 3106 | |
| Perodox 14 -40-PD | 2212-81-9 | CH ₃ | 80°C | | | 30°C | | Organic peroxide type D; solid | Division 4.1 | 1325 | |
| Perodox 14 -96-FL | 2212-81-9 | | 80°C | | | 20°C | | Organic peroxide type D; solid | Division 5.2 | 3106 | |
| Perodox 14 -96-PD | 25155-25-3 | | 80°C | | | 30°C | | Organic peroxide type D; solid | Division 5.2 | 3106 | |
| Perodox 14 -40-PD | 25155-25-3 | CH ₃ | 80°C | | | 30°C | | Organic peroxide type D; solid | Division 4.1 | | |
| Perodox 14 -40-GR | 25155-25-3 | | 80°C | | | 30°C | | Organic peroxide type D; solid | Division 4.1 | 1325 | |
| Perodox LUNA | 105-74-8 | O O | 50°C | | | 30°C | | Organic peroxide type D; solid | Division 5.2 | 3106 | |



Our Peroxide Products

| Common | | Molecular formula | Storage Data | | | | | Class/ | | | |
|-----------------|------------|---|--------------|---|--------------------------------|---------------|------------|---------------------------------|--------------|--------|-------|
| Name | CAS | | SADT /°C | Emergency temperature (T _e) | Control temperature (Tc) | Ts Max /°C | Ts Min /°C | Description | Divison | UN No. | PG |
| Perodox MEKP | 1338-23-4 | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 60°C | | | 25°C | | Organic peroxide type D; liquid | Division 5.2 | 3105 | PG II |
| TBEC | 34443-12-4 | $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 60°C | | | 20°C | | Organic peroxide type D; liquid | Division 5.2 | 3105 | |
| Perodox 42 | 13122-18-4 | CH ₃ O CH ₃ CH ₃ -C-CH ₂ -CH-CH ₂ -C-O-O-C-CH ₃ CH ₃ CH ₃ CH ₃ | 55°C | | | 25°C | -20°C | Organic peroxide type D; liquid | Division 5.2 | 3105 | |
| Perodox 101 | 78-63-7 | CH ₃ | 80°C | | | 40°C | 10°C | Organic peroxide type C; liquid | Division 5.2 | 3103 | |
| Perodox D24 | 133-14-2 | CI———————————————————————————————————— | 60°C | 30°C | ₁₅ ℃ | 30°C | | Organic peroxide type C; solid | Division 5.2 | 3104 | |





Your Safety is our Priority

In general, organic peroxides are thermally unstable components which can decompose at relatively low temperatures. However, knowledge of proper handling techniques, carefully designed facilities and thorough training of personnel can overcome the hazards. Personnel who understand and pay proper attention will be better able to handle organic peroxides confidently and safely.

UN numbers

Temperature

All products accepted for transport are assigned to generic entry numbers according to classification principles as described in the recommendations by the United Nations Committee of Experts on the Transport of Dangerous Goods. An explanation of all relevant UN numbers is given in Table 1.

Storage temperatures SADT:Self-Accelerating Decomposition

The SADT is the lowest temperature at which self-accelerating decomposition may occur with a substance in the packaging as used in transport. Transportation temperatures are derived from the SADT according to the recommendations by the United Nations Committee of Experts on the Transport of Dangerous Goods.

T_c max.

The T_s max. given in the product list on pages 8-10 is the recommended maximum storage temperature at which the product is stable and quality loss will be minimal.

T_c min.

Aminimum storage temperature (T_s min.) is given if phase separation, crystallization or solidification of the product is known to occur below the temperature indicated. We recommend that you store the product above the T_s min. indicated for quality and in some cases safety reasons.

T_{em}: Emergency temperature

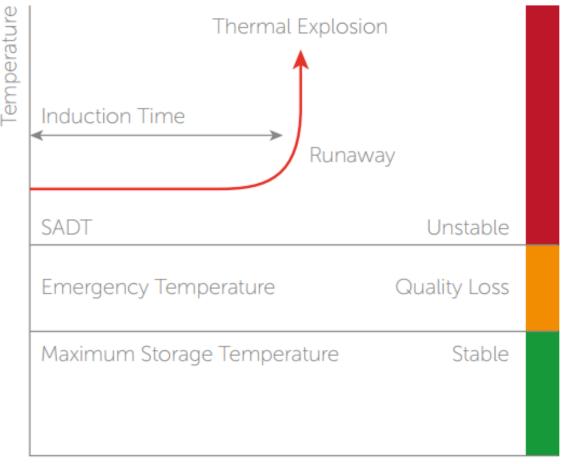
The $T_{\rm em}$ is derived from the SADT and is the temperature at which emergency procedures must be triggered.

T_c: Control temperatures

The T_c is also derived from the SADT and is the maximum temperature at which the product can be safely transported.

Both the T_{em} and T_c are related to safety and do not relate to product quality. To maintain product quality therecommended storage temperatures (T_s min. and max.) should be observed.

Survey of thermal stability



Time

Classification of organic peroxides:

| Divison | UN No. | Classification | Hazard Rating |
|--------------|--------|--|------------------|
| Division 5.2 | 3103 | type C; liquid | High |
| Division 5.2 | 3104 | type C; solid | High |
| Division 5.2 | 3105 | type D; liquid | Medium |
| Division 5.2 | 3106 | type D; liquid | Medium |
| Division 5.2 | 3107 | type E; liquid | Low |
| Division 5.2 | 3108 | type E; solid | Low |
| Division 5.2 | 3109 | type F; liquid | Very Low |
| Division 5.2 | 3110 | type F; solid | Very Low |
| Division 5.2 | 3113 | type C; liquid; temperature controlled | High |
| Division 5.2 | 3114 | type C; solid; temperature controlled | High |
| | | | |



Safety Aspects

We are recognized as the global leader in organic peroxide safety. Our proven success in safety handling organic peroxides is due to our long-term commitment to developing and maintaining high safety standards. We always place safety as our top priority.

In general, organic peroxides are unstable chemical compounds, decomposing at relatively low temperatures. They can, however, be handled and stored safely if proper precautions are followed. The prevention of undesired decomposition reactions during handling and storage of organic peroxides require knowledge of the general properties of organic peroxides and the specific requirements of each individual peroxide. Personnel who understand and pay proper attention will be able to handle organic peroxides confidently and safely

This brochure gives guidelines for the safe storage of organic peroxides in their original packaging. Organic peroxide storage requires two important considerations:

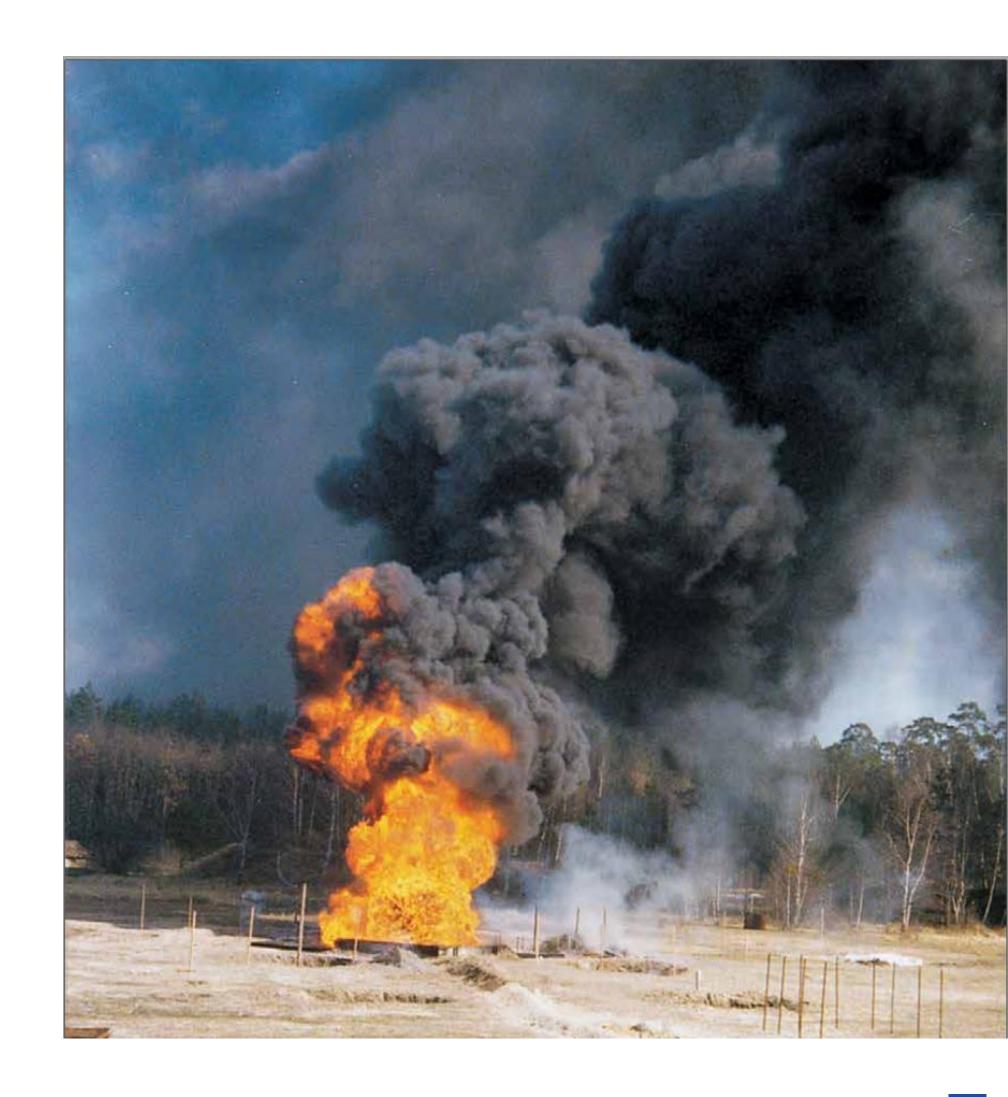
- minimization of the possibility of a peroxide decomposition
- reduction of the effects of such a decomposition.

This publication includes guidelines to minimize the probability of a peroxide decomposition by limiting the heat exposure of the product and preventing contamination. Measures to reduce effects of a decomposition are also discussed.

General recommendations are given as well as specific measures for temperature-controlled products.

Beyond the scope of this brochure are the local laws and insurance regulations that must be considered in the design of storage facilities. Various countries have published directives with their local laws.

For more information on the use and the safe handling and storage of organic peroxides, please contact your Do Sender Chem account manager or regional sales office.



Fire of 5,000 kgs organic peroxide



Safety Aspects

For product inquiry and ordering information, please contact your Do Sender Chemical account manager or regional sales office.

Thermal sensitivity

Organic peroxides are thermally unstable due to the relatively weak
-O-O- bond in the molecular structure. As a result of this property organic peroxides are sensitive to heat and will decompose significantly above a certain temperature. This temperature is specific to each individual peroxide.

Every packaged peroxide has a characteristic temperature at which the self-heating accelerates. This temperature is designated as the Self-Accelerating Decomposition Temperature (SADT). At or above this temperature a runaway reaction will take place. For this reason, strict temperature control and high temperature alarms are required for the storage of organic peroxides. For many peroxides refrigerated facilities will be necessary.

During decomposition heat is released and the rate of this decomposition increases with temperature. If this heat cannot be transferred to the environ- ment due to reduced surface area from stacking or high ambient temperatures, self-heating will accelerate and lead to a violent combustion or thermal explosion.



Safety testing to determine the sensitiveness of peroxides to the effect of heat under confinement



Contamination

Contamination may accelerate the decomposition of organic peroxides.

Most heavy metal compounds have an accelerating effect on decomposition.

Acids, bases and accelerators based on e.g. cobalt or copper and amines may cause decompositions to occur at temperatures significantly lower than the recommended storage temperature.

Therefore, organic peroxides must be stored separately from other chemical compounds, unless compatibility has been proven.

Burning properties

Most organic peroxides should be considered highly combustible. Once ignited many organic peroxides burn vigorously. When heated to their decomposition temperature, organic peroxides will generate vapors.

Many of these vapors are flammable.

Therefore, the presence of potential ignition sources must be avoided.

Electrical equipment installed should be explosion-proof to avoid sparks.

Nevertheless, auto-ignition of the vapors may still occur. Consequently, fire-fighting equipment should be present in all storage rooms.

Pressure build-up

As organic peroxides produce vapors during decomposition, pressure build-up in the package occurs. To prevent excessive pressure rise inside storage rooms due to prolonged decomposition and bursting of the packages, the storage room should be provided with a pressure relief mechanism.



Storage

One can minimize the probability of a peroxide decomposition by limiting the heat exposure of the product and preventing contamination. Temperature control is the most critical control measure in preserving quality and preventing a runaway reaction

Although a number of organic perox- ides can be safely stored at ambient temperature, most require some form of temperature control. The maximum temperature allowed by the regulatory agencies is the control temperature T_c .

This temperature together with the emergency temperature ($T_{em'}$ see section Temperature control and monitoring on this page) are derived from the SADT in next table.



| TYPE OF RECEPTACLE | SADT | CONTROL TEMPERATURE | EMERGENCY TEMPERATURE |
|-------------------------------|--|---|--|
| Single packagings and IBCs | 20°C or less over 20 to 35°C over 35°C | 20°C below SADT 15°C below SADT 10°C below SADT | 10°C below SADT 10°C below SADT 5°C below SADT |
| Portable tanks | ≤45°C | 10°C below SADT | 5°C below SADT |

However, for a longer shelf life, lower storage temperatures than the control temperature are generally recommended. At this recommended storage temperature (T_s), as indicated on the product label, the product will be stable and quality loss will be minimal.

As each organic peroxide has its own specific storage temperature, consult the product catalog or individual Product Data Sheet (PDS) which can be downloaded from polymerchemistry. nouryon.com.

Cooled storage rooms should be provided with at least two independent temperature alarms. An alarm is recommended when the storage temperature is exceeded by 5°C. If this is the case the store should be inspected. The temperature alarm should incorporate some delay device to allow for intermittent short temperature increase which result from inspection, loading and unloading, etc. Any possibility of alarm de-activation should be countered with an automatic alarm re-activation.

Temperature monitoring and control

The temperature must be maintained at or below the recommended storage temperature; consult the product catalog or PDS.

Organic peroxides should be protected from direct sunlight and all other sources of heat.

Particular attention should be given to:

- windows (not recommended or should be blinded),
- heating systems (e.g. radiators, warmwater pipes, etc).

Non-cooled storage rooms should be provided with an alarm set at the emergency temperature but not higher than 45°C. It is imperative that a signal from the mentioned alarm systems should not escape notice at any time including nights, weekends, etc. and that appropriately trained personnel are alerted.

There should be dual power supply for the alarm system.



Storage

Fire fighting

A deluge system of large capacity is strongly recommended to fightlarge fires.

A dry-powder fire extinguisher with a capacity of at least 10 kg should be located outside the storage building, near the entrance. This is used only to fight small fires of organic peroxides.

After the fire has been extinguished the peroxide should be cooled to a temperature below the SADT to prevent the peroxide from re-igniting due to self heating.



Smoking, open fire and all other sources of ignition must be forbidden in and near the storage rooms. Appropriate warnings should be posted in the storage area.

Unless compatibility has been proven, organic peroxides should be stored separately from other chemical compounds and, under all circum- stances, away from accelerators and other reducing agents.

Use vermiculite or perlite to clean up small spillages



Small fire extinguishers in the neighborhood of peroxide stores



Store organic peroxides in their original closed packages.

Packages should be visually checked for labelling, leakage and damage, etc. at the time of delivery. If necessary the material should be repacked in suitable, clean packages. A suitable packaging material is non-pigmented polyethylene. Never use tightly closed rigid metal containers.



Space between pallets to allow maximum air circulation

Repacking, weighing, mixing, etc. should be carried out in a separate room. Clean dedicated polyethylene or stainless steel implements should be used to prevent contamination of the peroxide.

Stacking procedures should require that labels and safety information are always visible.

Pallets should be configured to allow maximum air circulation, no more than two high with a minimum clearance of 0.1 m from anywall.

Storage on racks may be applied when individual packages are stored.

Different products should be stacked separately to avoid confusion.

If liquid organic peroxides are stored, an inert absorbent material, such as vermiculite or perlite, should be available in or near the store to clean up small spillages. Absorbents should be soaked with water afterwards.

Stock should be rotated following the first in, first out principle.

The doors of the storage should be marked with the peroxide label.

The storage building should bekept clean; no litter, rags, etc.

The store should be kept closed with access limited to authorised, trained personnel only.



Directives for the safe handling and storage of organic peroxides

Storage

Organic peroxides should be protected against all sources of heat, even direct sunlight. Storage together with other chemicals, especially accelerators, other reducing materials and inflammable products must be avoided.

Handling

Fire hazard No smoking, no naked lights, no sparks, or other sources of ignition

Explosion hazard

Avoid direct contact of organic peroxides with accelerators — add each component separately to the resin. Contamination with dust, heavy metals and their compounds, as well as chemicals in general, should be avoided.

Eye and skin injury

Always wear safety goggles and protective gloves, since organic peroxides have a corrosive effect on eyes and skin.

Additional information

On request we also provide specific publications on the use and the safe handling and storage of our products

How to act in case of:



Fire

Alert fire department. Fight small fire with powder or carbon dioxide and apply water.



Spillage

Liquids: absorb with inert material and add water.
Solids/pastes: take up with compatible aids and add water. Move to safe place and arrange disposal as soon as possible.



Skin contact

Wash with water and soap.



Eye contact

First rinse with water for at least 15 minutes. Always seek medical attention.



Ingestion

Drink large amounts of water and consult doctor immediately.
Do not induce vomiting.

Do Sender Chem is recognized as the global leader in organic peroxide safety. We always place safety as our top priority.

Sharing our experience in safety is one of the most important resources we offer Classroom reviews of safety and handling of organic peroxides, online trainings, consultation on storage and peroxide dosing equipment as well as demonstrations and publications on the safe use and handling of organic peroxides are just some of the services we offer.

How to store peroxides





- Store in a cool room away from direct sunlight.
- Observe maximum and minimum storage temperature as printed on the packagingand SDS.
- Leave in the original packaging.
- Close packaging after use.



- Do not store together with accelerators or other chemicals
- Do not mix peroxides with accelerators.
- Avoid any contact with dust, metal or other chemicals.

How to handle peroxides





- Wear safety goggles.
- Wear appropriate protective gloves and clothing.
- Remove spillages immediately.
- Only use compatible materials when handling.



- Do not smoke.
- Avoid heat sources
- Avoid open fire.
- Never heat peroxides



Packaging

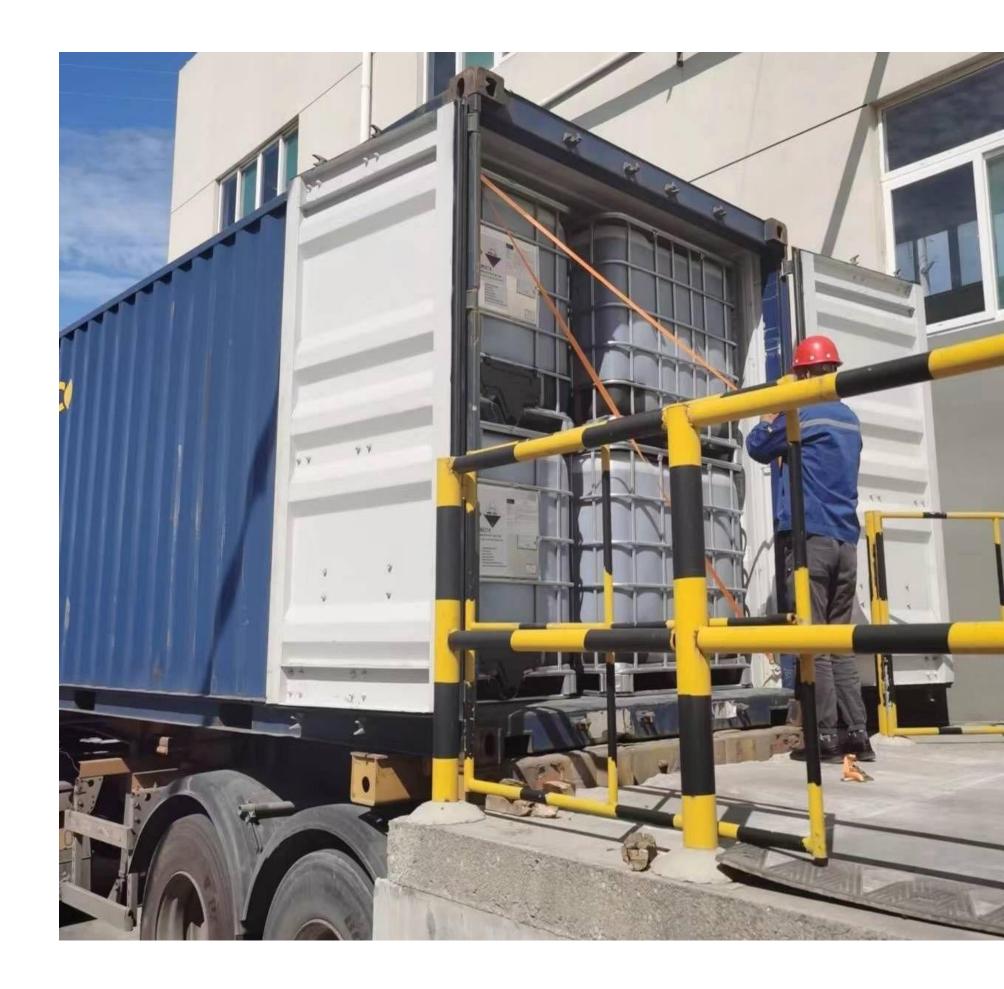
We continuously develop new and innovative packaging making logistics more efficient and improving safety standards even beyond existing transport regulations for both liquid and solid organic peroxides.



Do Sender Chem offers a variety of packaging options for both liquid and solid organic peroxides. For details of the packaging pls visit us at www.perodox.com or contact our sale for further information.

Enhanced advantages and safety features

- Easy operating, screw cap anti-glug device
- Ergonomically designed handle for ease and safety of handling
- Opaque exterior to protect contents from harmful UV rays
- Shaped for promotion of optimal air circulation while stacked
- Unique interior features allow more complete drainage







Perodox B

CAS No.

110-05-4

TSCA Status

listed on inventory

Active oxygen content peroxide

10.94%

EINECS/ELINCS No.

203-733-6

Molecular weight

146.2

Characteristics

Clear liquid

Density, 20 °C

 $0.800 \, g/cm^3$

Viscosity, 20 °C

0.9 mPa.s

$$CH_3$$
 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

Di-Tert-Butyl Peroxide

Polymerization of acrylate and methacrylate: Within the temperature range of 130-175 °C, it can be used as an initiator for solution polymerization or copolymerization of acrylate and methacrylate, especially in the production of coatings. It can also be used as an initiator for bulk and suspension polymerization or copolymerization of acrylate and methacrylate.

Applications

DTBP(Di-tert-butyl peroxide) can be used for the market segments: polymer production, polymer crosslinking and acrylics production with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t½) at various temperatures. For DTBP in chlorobenzene half-life at other temperatures can be calculated by using the equations and constants mentioned below:

0.1 hr at 164°C (327°F) at 141°C (286°F) 1 hr at 121°C (250°F) 10 hr $kd = A \cdot e - Ea/RT$ Formula 1 Formula 2 $t\frac{1}{2} = (\ln 2)/kd$ 153.46 kJ/mole Ea Α 4.20E+15 s-1 R 8.3142 J/mole·K (273.15+°C) K

Thermal stability

Organic peroxides are thermally unstable substances which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition may occur with a substance in the packaging as used for transport is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $80^{\circ}\text{C} (176^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides, a loss of quality will occur over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 40° C (104° F) and

Ts Min. -30°C (-22°F) to prevent crystallization

Note When stored according to these recommended storage

conditions, DTBP will remain within the Do Sender Chem specifications for a period of at least 6 months after

delivery.

Packaging and transport

Polyethylene white plastic drum 20KG.

DTBP is classified as Organic peroxide type E; liquid, Division 5.2; UN 3107.

Major decomposition products

Acetone, Methane, tert-Butanol.



Perodox K90

CAS No.

80-15-9

TSCA Status

listed on inventory

EINECS/ELINCS No.

210-254-7

Molecular weight

152.2

Appearance

Clear liquid

Density, 20 °C

 $1.040 \, \text{g/cm}^3$

Viscosity, 20 °C

5 mPa.s

Cumyl Hydroperoxide 90%

Perodox K90 is an initiator (90% active ingredient in aromatic solvent mixture) for (co)polymerization of (meth)acrylates.

Applications

For Polymer production and Poly(meth)acrylics: Perodox K90 may be used for various polymerization reactions. It can be used in emulsion, solution and bulk polymerizations. In emulsion processes, CHP may be activated by organic-soluble or water-soluble reducing agents, or by metal compounds to achieve polymerization at room temperature or lower. When no accelerators are used, effective polymerization can be obtained in the temperature range of 50-200°C. For example, styrene and methyl methacrylate can be polymerized in bulk in the temperature range of 60-100°C using CHP. CHP may also be used for emulsion polymerization of various vinyl monomers. In this case CHP may be used in combination with reducing agents to achieve reproducible results at low temperatures. For Thermoset: CHP may be used as an initiator for the room temperature cure of promoted unsaturated polyester and vinyl ester resins, and elevated temperature cure of non-promoted resins.

Thermal stability

Organic peroxides are thermally unstable substances which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition may occur with a substance in the packaging as used for transport is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 60°C

Method The Heat Accumulation Storage Test is a recognized test

method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria -

United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature

Ts Max. 25° C

Note When stored under the recommended storage

conditions, Perodox K90 will remain within the Do Sender Chem specifications for a period of at least

9 months after delivery.

Packaging and transport

Packed in plastic drums with specifications of 1000kg, 200kg, and 25kg. Perodox K90 is classified as Organic peroxide type F; liquid, Division 5.2; UN 3109.

Major decomposition products

Acetophenone, phenylisopropanol, methane, water.



Perodox K80

CAS No.

80-15-9

TSCA Status

listed on inventory EINECS/ELINCS No.

210-254-7

Molecular weight

152.2

Appearance

Clear liquid

Density, 20 °C

 $1.06 \, \text{g/cm}^3$

Viscosity, 20 °C

10.4 mPa.s

Cumyl Hydroperoxide 80%

Perodox K80 may be used for various polymerization reactions. It can be used in emulsion, solution and bulk polymerizations. In emulsion processes, CHP80 may be activated by organic-soluble or water-soluble reducing agents, or by metal compounds to achieve polymerization at room temperature or lower. When no accelerators are used, effective polymerization can be obtained in the temperature range of 50-200°C. For example, styrene and methyl methacrylate can be polymerized in bulk in the temperature range of 60-100°C using Perodox K80 . Perodox K80 may also be used for emulsion polymerization of various vinyl monomers. In this case CHP90 may be used in combination with reducing agents to achieve reproducible results at low temperatures.

Applications

Perodox K80 can be used for the market segments: polymer production, thermoset composites and acrylics production with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t1/2) at various temperatures. The half-life of CHP80 in chlorobenzene is:

0.1 hr at 195°C

1 hr at 166°C

10 hr at 140°C

Formula 1 kd=A·e-Ea/RT

Formula 2 t½=(ln2)/kd

Ea 132.56 kJ/mole

A 1.15E+12 s-1

R 8.3142 J/mole·K T (273.15+°C) K

Thermal stability

Organic peroxides are thermally unstable substances which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition may occur with a substance in the packaging as used for transport is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 75°C for small cans, 70°C for IBC's and 65°C for bulk tanks.

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature

Ts Max. 40° C Ts Mix. -30° C *

Note When stored under the recommended storage conditions, CHP80 will remain within the Do Sender Chem specifications for a period of at least 6

months after delivery.

Packaging and transport

Packed in plastic drums with specifications of 1000kg, 200kg, and 25kg. CHP is classified as Organic peroxide type F; liquid, Division 5.2; UN 3109.

Major decomposition products

Acetophenone, 2-Phenylisopropanol, Methane



Perodox L75

CAS No.

94-36-0

TSCA Status

listed on inventory EINECS/ELINCS No.

202-327-6

Molecular weight

242.2

Active oxygen

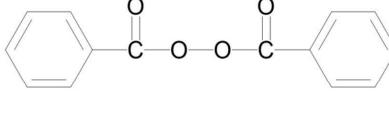
4.88-5.02 %

Appearance

White granular powder

Assay

74.0-76.0 %



(Di)Benzoyl peroxide 75%

Perodox L75 is an initiator (powder formulation with 75% benzoyl peroxide and 25% water) used for curing unsaturated polyester, vinyl ester and acrylic thermoset resins at ambient or slightly elevated temperatures. It is often used in conjuction with tertiary amine accelerators at ambient conditions. Typical uses include cast polymer, panels, chemical anchors and mine bolts and RTM.

Applications

BPO75 can be used for the market segments: polymer production, thermoset composites and acrylics production with their different applications/functions.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t1/2) at various temperatures. For BPO75 in chlorobenzene half-life at other temperatures can be calculated by using the equations and constants mentioned below:

| 0.1 hr | at 113°C (235°F) |
|-----------|--------------------------|
| 1 hr | at 91°C (196°F) |
| 10 hr | at 71°C (160°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 122.35 kJ/mole |
| A | 6.94E+13 s-1 |
| R | 8.3142 J/mole∙K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 80°C Emergency temperature (Te) 75°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide.

Ts Max. 40° C

Note When stored under the recommended storage conditions,

BPO75 will remain within the Do Sender Chem

specifications for a period of at least 3 months after

delivery.

Packaging and transport

Packed in plastic drums with specifications of 25kg. BPO75 is classified as Organic peroxide type C; solid, Division 5.2; UN 3104; PG II.

Major decomposition products

Carbon dioxide, Benzene, Benzoic acid



Perodox L50-PS

CAS No.

94-36-0

TSCA Status

listed on inventory

EINECS/ELINCS No.

202-327-6

Molecular weight

242.2

Active oxygen content peroxide

6.61%

Concentration

3.24-3.37%

(Di)Benzoyl peroxide 50%

Perodox L50-PS is the preferred BPO paste for unsaturated polyester & vinyl ester resins.

Applications

Perodox L50-PS is a paste containing 50% dibenzoyl peroxide without phthalate for the curing of unsaturated polyester resins at ambient and elevated temperatures. At temperatures up to 80°C, BPO50 Paste should be used in combination with an aromatic tertiary amine accelerator, above 80°C the use of an accelerator is not required. BPO50 Paste has primarily been developed for the putty market. BPO50 Paste shows a very good chemical and physical stability and is therefore very suitable for tube filling.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $70^{\circ} \text{C} (158^{\circ} \text{F})$

Method The Heat Accumulation Storage Test is a recognized test

method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United

Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide.

Ts Max. 30° C (86° F)

Note When stored under the recommended storage conditions,

BPO50 Paste will remain within the Do Sender Chem specifications for a period of at least 3 months after

delivery.

Packaging and transport

Packed in plastic drums with specifications of 25kg. BPO50 Paste is classified as Organic peroxide type E; solid, Division 5.2; UN 3108; PG II.

Major decomposition products

Carbon dioxide, benzene, benzoic acid



Perodox C

CAS No.

614-45-9

TSCA Status

listed on inventory EINECS/ELINCS No.

210-382-2

Molecular weight

194.2

Active oxygen

8.07-8.24 %

Appearance

Clear liquid

Assay

≥ 98.0 %

Tert-Butyl peroxy benzoate

In the temperature range of 100-170°C, TBPB can be used as an initiator for the solution polymerization or copolymerization of acrylate and methacrylate, especially for the production of coatings.

Applications

TBPB can be used for the market segments: polymer production, polymer crosslinking, thermoset composites and acrylics production with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t½) at various temperatures. TBPB in chlorobenzene half-life at other temperatures can be calculated by using the equations and constants mentioned below:

0.1 hr at 142°C (288°F) at 122°C (252°F) 1 hr at 103°C (217°F) 10 hr Formula 1 $kd = A \cdot e - Ea/RT$ $t^{1/2} = (\ln 2)/kd$ Formula 2 151.59 kJ/mole Ea 2.23E+16 s-1 8.3142 J/mole·K R $(273.15 + {}^{\circ}C) K$ Т

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $60^{\circ}\text{C} (140^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Note

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 25° C (77°F) Ts Min. 10° C (50°F)*

* to prevent crystallization. When stored under the recommended storage conditions, Do Sender Chem will remain within the specifications for a period of at least 3 months after delivery.

Packaging and transport

25 kg polyethylene packaging TBPB is classified as Organic peroxide type C; liquid; Division 5.2; UN 3103.

Major decomposition products

Carbon dioxide, Acetone, Methane, tert-Butanol, Benzoic acid, Benzene



CAS No.

3006-82-4 TSCA Status

listed on inventory

Active oxygen content peroxide

7.40%

EINECS/ELINCS No.

221-110-7

Molecular weight

216.3

Concentration

7.22-7.37%

$$CH_3$$
— $(CH_2)_3$ — CH — C — O — O — C — CH_3
 C_2H_5
 CH_3

Tert-butyl peroxy-2-ethylhexanoate

Perodox 21 is an efficient initiator for the production of Low Density Polyethylene (LDPE). It is used both for tubular and autoclave processes. In most cases a combination with other peroxides is used to ensure a broad reactivity range.

Applications

TBPEH(tert-Butyl peroxy-2-ethylhexanoate) can be used for the market segments: polymer production, thermoset composites and acrylics with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t1/2) at various temperatures. For TBPEH in chlorobenzene half-life at other temperatures can be calculated by using the equations and constants mentioned below:

0.1 hr at 113°C 1 hr at 91°C 10 hr at 72°C

Formula 1 kd = A·e-Ea/RT Formula 2 t¹/2 = (ln2)/kd Ea 124.90 kJ/mole A 1.54E+14 s-1 R 8.3142 J/mole·K T (273.15+°C) K

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 35°C Emergency temperature (Te) 25°C Control temperature (Tc) 20°C

Method

The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 10°C and

Ts Min. -30°C to prevent crystallization

Note When stored according to these recommended storage conditions, TBPEH will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg polyethylene drum.TBPEH is classified as Organic peroxide type C; liquid, temperature controlled, Division 5.2; UN 3113; PG II. Control Temperature = 20 °C Emergency Temperature = 25 °C

Major decomposition products

Carbon dioxide, tert-Butanol, Heptane, 3-tert-Butoxyheptane



CAS No.

25155-25-3,2212-81-9

TSCA Status

listed on inventory

Active oxygen content peroxide

9.45%

EINECS/ELINCS No.

218-664-7

Molecular weight

338.5

Concentration

9.08% min.

1,3-Di-(2-tert-butylperoxyisopropyl)benzene

1,4-Di-(2-tert-butylperoxyisopropyl)benzene

Perodox 14 is a bifunctional peroxide which is used for the crosslinking of natural rubber and synthetic rubbers, as well as polyolefins. Rubber compounds containing BIPB have excellent scorch safety, and under certain conditions one step mixing is possible. Safe processing temperature: 135°C (rheometer ts2 > 20 min.). Typical crosslinking temperature: 175°C (rheometer t90 about 12 min.).

Applications

Perodox 14 can be used for the market segments: polymer production and polymer crosslinking with their different applications/functions.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $80^{\circ}\text{C} (176^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. $30^{\circ} \text{C} (86^{\circ} \text{F})$

Note Perodox 14 can be safely stored at 30°C (86°F) max without loss of activity. When stored under strictly recommended storage conditions, BIPB will remain within the Do Sender Chem specifications for a period of at least 12 months after delivery.

Packaging and transport

20 kg corrugated box packaging BIPB is classified as Organic peroxide type D; solid, Division 5.2; UN 3106. PG II

Major decomposition products

tert-Butanol, Methane, Acetone, Bis(2-hydroxyisopropyl)benzene.

Attn

Perodox 14-96% owns flakes and powder forms Perodox 14-40% owns powder and granule forms Usually Perodox 14 combined with CAS 1025-15-6 TAIC Triallyl isocyanurate as the Crosslinking Coagent



Perodox LUNA

CAS No.

105-74-8

TSCA Status

listed on inventory

Active oxygen content peroxide

4.01%

EINECS/ELINCS No.

203-326-3

Molecular weight

398.6

Appearance

White flakes without any contamination

Assay

≥ 99.0 %

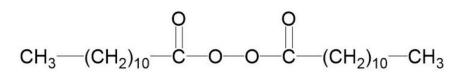
Perodox LUNA is a widely used initiator for the suspension and mass polymerization of vinyl chloride between 60°C and 80°C. In many cases LPO is combined with a more active initiator, such as a peroxydicarbonate to increase reactor efficiency. LPO is used as an initiator for the high pressure polymerization of ethylene, but because of its poor solubility in most aliphatics, it is in many cases replaced by other peroxides such as Di(3,5,5-trimethylhexanoyl) peroxide (TMHP). The advantage of LPO is the possibility of storing at ambient temperature. LPO is also used as an initiator for the polymerization of methylmethacrylate at 60-90°C. LPO is often applied as a replacement for 2,2'-Azobis(isobutyronitril) (AIBN).

Applications

LPO can be used for the market segments: polymer production, thermoset composites and acrylics production with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t1/2) at various temperatures. For LPO in chlorobenzene:



Di lauroyl peroxide

Thormal stability

at 99°C

at 79°C

at 61°C

 $kd = A \cdot e - Ea/RT$

 $t^{1/2} = (\ln 2)/kd$

123.37 kJ/mole

3.92E+14 sP-1P

8.3142 J/mole·K

 $(273.15 + {}^{\circ}C) K$

0.1 hr

1 hr

10 hr

Ea

A

R

Τ

Formula 1

Formula 2

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 50°C

Method The Heat Accumulation Storage Test is a recognized test method for thedetermination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva)

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30°C

Note When stored under these recommended storage conditions, LPO will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg corrugated box packaging LPO is classified as Organic peroxide type D; solid, Division 5.2; UN 3106. PG II

Major decomposition products

Carbon dioxide, Docosane, Undecane, Undecyl dodecanoate.



Perodox MEKP

CAS No.

1338-23-4

TSCA Status

listed on inventory

EINECS/ELINCS No.

215-661-2

Appearance Clear colorless liquid

Total active oxygen

8.8-9.0 %

Methyl Ethyl ketone Peroxide

Applications

Perodox MEKP is a general purpose methyl ethyl ketone peroxide (MEKP) for the curing of unsaturated polyester resins in the presence of a cobalt accelerator at room and elevated temperatures. The curing system Perodox MEKP-50/cobalt accelerator is particularly suitable for the curing of gelcoat resins, laminating resins, lacquers and castings; moreover the manufacture of light resistant parts may be possible contrary to the curing system benzoyl peroxide/amine accelerator. Practical experience throughout many years has proven that by the guaranteed low water content and the absence of polar compounds in Perodox MEKP-50, this peroxide is very suitable in GRP products for e.g. marine applications.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 60°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 25° C

Note When stored under these recommended storage conditions, MEKP will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg corrugated box packaging.

Perodox MEKP is classified as Organic peroxide type D; liquid, Division 5.2; UN 3105.

Major decomposition products

Carbon dioxide, water, acetic acid, formic acid, propionic acid, methyl ethyl ketone.

Attn

Based on the different active oxygen of the MEKP series, Do Sender Chem makes a types table, Please contact us for advice on the best curing system for your specific application



Perodox MEKP-Types of Active Oxygen

| DDODLICT | DECCDIDETON | ACTIVE OXYGEN | SAFETY INFORMATION | | |
|---|--|---------------|-----------------------|-----------|--|
| PRODUCT | DESCRIPTION | CONTENT | Ts (°C) | SADT (°C) | |
| | Methyl Ethyl Ketone Peroxide CAS 1338-23-4 | | | | |
| Perodox MEKP-10 | Standard, general purpose MEKP, low water content, absence of polar solvents | 9.9 | 25 | 60 | |
| Perodox MEKP-90 | Standard, general purpose MEKP, low water content, absence of polar solvents | 8.9 | 25 | 60 | |
| Perodox MEKP-90H | Fast gel time, general purpose lamination | 9.9 | 25 | 60 | |
| Perodox MEKP-90L | High Dimer, designed for VE resins & gel coats. Less foaming | 8.5 | 25 | 60 | |
| Perodox MEKP-100 | Economical, general purpose MEKP | 8.9 | 25 | 55 | |
| Perodox MEKP-200 | Economical, general purpose MEKP | 9.9 | 25 | 55 | |
| Perodox MEKP-1000P | MEKP gel, designed for putty curing | 8 | 25 | 55 | |
| Methyl Ethyl Ketone Peroxide (Phthalate Free) CAS 1338-23-4 | | | | | |
| Perodox MEKP-90A | Standard, general purpose MEKP, low water content, Phthalate free | 8.9 | 25 | 60 | |
| Perodox MEKP-90HA | Fast gel time, general purpose lamination, Phthalate free | 9.9 | 25 | 60 | |



CAS No.

34443-12-4 TSCA Status CH_3 — CH_2 — CH_2 — CH_2 — CH_3 — CH_3 — CH_3 — CH_4 — CH_5 — CH_3

EINECS/ELINCS No.

252-029-5 Appearance tert-Butylperoxy 2-ethylhexyl carbonate

Clear colorless liquid

Active oxygen content peroxide

6.49%

Appearance

Clear liquid

Assay

≥ 95.0 %

Applications

Perodox 117 can be used as initiator for the solution (co)polymerization of acrylates and methacrylates in the temperature range of 100-170°C, amongst others for the manufacture of coatings. It can also be applied as an initiator for the bulk and suspension (co)polymerization of acrylates and methacrylates.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t^{1/2}$) at various temperatures. For Perodox 117 in chlorobenzene:

| 0.1 hr | at 137°C (279°F |
|-----------|------------------------|
| 1 hr | at 117°C (243°F |
| 10 hr | at 98°C (208°F) |
| Formula 1 | kd = A·e-Ea/RT |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 151.72 kJ/mole |
| A | 4.07E+16 s-1 |
| R | 8.3142 J/mole·K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $60^{\circ}\text{C} (140^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. $20^{\circ}\text{C}(68^{\circ}\text{F})$

Note When stored under these recommended storage conditions, MEKP will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

25 kg corrugated box packaging.

Perodox 117 is classified as Organic peroxide type D; liquid, Division 5. 2; UN 3105.

Major decomposition products

Carbon dioxide, tert-Butanol, 2-Ethylhexanol



CAS No.

13122-18-4 TSCA Status

listed on inventory

EINECS/ELINCS No.

236-050-7

Active oxygen

6.74 %

Appearance

Clear liquid

Assay

97%

Color 50 Pt-Co / APHA max.

tert-Butyl peroxy-3,5,5-trimethylhexanoate

Perodox 42S is an initiator for (co)polymerization of (meth)acrylates.

Applications

Perodox 42 (tert-Butyl peroxy-3,5,5-trimethylhexanoate) can be used for the market segments: polymer production, thermoset composites and acrylics with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 42 in chlorobenzene:

135°C (275°F) 0.1 hr 114°C (237°F) 1 hr 94°C (201°F) 10 hr Formula 1 $kd = A \cdot e - Ea/RT$ $t^{1/2} = (\ln 2)/kd$ Formula 2 140.78 kJ/mole Ea 1.94E+15 s-1 Α 8.3142 J/mole·K (273.15+°C) K

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $55^{\circ}C (131^{\circ}F)$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see

Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and

Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 25° C (77°F)

Note When stored under these recommended storage conditions,

Perodox 42 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg. Perodox 42 is classified as Organic peroxide type D; liquid, Division 5.2; UN 3105.

Major decomposition products

Carbon dioxide, Methane, tert-Butanol, Acetone, 2-tert-Butyloxy-2,4,4-trimethylpentane



CAS No.

78-63-7

TSCA Status

listed on inventory

EINECS/ELINCS No.

201-128-1

Active oxygen

≥ 10.14 % 2,5-Dimethyl-2,5-di(tert-butylperoxy) hexane

Appearance

Clear liquid

Assay

≥ 92.0 %

Color

 \leq 50 Pt-Co / APHA

Applications

Perodox 101 can be used for the market segments: polymer production, polymer crosslinking, acrylics production and polymer recycling with their different applications/functions. For more information please check our website and/or contact us.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 101 in chlorobenzene:

| 0.1 hr | at 156°C (313°F) |
|-----------|--------------------------|
| 1 hr | at 134°C (273°F) |
| 10 hr | at 115°C (239°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 155.49 kJ/mole |
| A | 1.68E+16 s-1 |
| R | 8.3142 J/mole∙K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self-Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $80^{\circ}\text{C} (176^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 40° C (104° F) and Ts Min. 10° C (50° F)

Note When stored under these recommended storage conditions,

Perodox 101 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg

Perodox 101 is classified as Organic peroxide type C; liquid, Division 5.2; UN 3103.

Major decomposition products

Acetone, Methane, tert-Amyl alcohol, tert-Butanol, Ethane



Perodox D24

CAS No.

133-14-2

TSCA Status

listed on inventory EINECS/ELINCS No.

205-094-9

Molecular weight

380.0

Active oxygen content peroxide

4.21%

Appearance

Off-white homogeneous paste

Assay

49.0-51.0 %

Concentration

2.06-2.15%

Di(2,4-dichlorobenzoyl) peroxide

Applications

Perodox D24 is a monofunctional peroxide which is used for the crosslinking of silicone rubber. Safe processing temperature: 75°C (rheometer ts2 > 20 min.). Typical crosslinking temperature: 90°C (rheometer t90 about 12 min.).

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $60^{\circ}\text{C} (140^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30° C (86°F) Ts Min. 10° C (50°F)

Note When stored under these recommended storage conditions,

Perodox D24 will remain within the Do Sender Chem

specifications for a period of at least 6 months after delivery.

Packaging and transport

20 kg.

Perodox D24 is classified as Organic peroxide type C; solid, Division 5.2; UN 3104.

Major decomposition products

Carbon dioxide, 1,3-Dichlorobenzene, 2,4-Dichlorobenzoic acid, Traces of 2,2',4,4' tetrachlorobiphenyl



CAS No.

4511-39-1

TSCA Status

listed on inventory EINECS/ELINCS No.

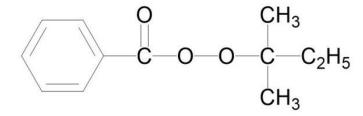
224-831-5

Molecular weight

208.3

Active oxygen content peroxide

7.68%



Tert-Amyl peroxybenzoate

Applications

Polymerization of styrene Perodox 127 may be used for the (co)polymerization of styrene in the temperature range of 100-135°C. In practice, combinations of two or more peroxides with diverging activities are used to reduce the residual monomer content in the final polymer and to increase reactor efficiency. During polymerization, the temperature is increased incrementally to ensure that the temperature necessary to attain the optimum properties for each peroxide is reached. In this respect, Perodox 127 is often used with more active initiators, such as dibenzoyl peroxide (Perodox L75) or tert-butyl peroxy-2-ethylhexanoate (Perodox 21). In a mass process Perodox 127 may be used to increase the rate of polymerization.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t^{1/2}$) at various temperatures. For Perodox 127 in chlorobenzene:

| 0.1 hr | at 139°C (282°F) |
|-----------|----------------------------|
| 1 hr | at 118°C (244°F) |
| 10 hr | at 99°C (210°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 147.02 kJ/mole |
| A | 8.38E+15 s-1 |
| R | 8.3142 J/mole·K |
| T | $(273.15 + {}^{\circ}C) K$ |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $60^{\circ}\text{C} (140^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 20° C (68°F)

Note When stored under these recommended storage conditions,

Perodox 127 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 127 is classified as Organic peroxide type C; liquid; Division 5. 2; UN 3103.

Major decomposition products

Carbon dioxide, Benzene, Benzoic acid, tert-Amylalcohol, Acetone, Ethane



Perodox F50

CAS No.

107-71-1

TSCA Status

listed on inventory EINECS/ELINCS No.

203-514-5

Molecular weight

132.2

Active oxygen content peroxide

12.11%

$$CH_3$$
 CH_3 CH_3 CH_3 CH_3 CH_3

tert-Butyl peroxyacetate, 50% solution

Applications

Perodox F50 can be used as initiator for the solution (co)polymerization of acrylates and methacrylates in the temperature range of 100-170°C, amongst others for the manufacture of coatings. Perodox F50 can also be applied as an initiator for the bulk and suspension (co)polymerization of acrylates and methacrylates.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t^{1/2}$) at various temperatures. For Perodox F50 in chlorobenzene:

0.1 hr at 139°C (282°F) at 119°C (246°F) 1 hr at 100°C (212°F) 10 hr $kd = A \cdot e - Ea/RT$ Formula 1 $t^{1/2} = (\ln 2)/kd$ Formula 2 149.36 kJ/mole Ea 1.57E+16 s-1 Α 8.3142 J/mole·K $(273.15 + {}^{\circ}C) K$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $70^{\circ}\text{C} (158^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 10° C (50° F)

Ts Min. -15° C (5° F) to prevent phase separation

Note When stored under these recommended storage conditions,

Perodox F50 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox F50 is classified as Organic peroxide type C; liquid, Division 5. 2; UN 3103

Major decomposition products

Carbon dioxide, Acetone, Methane, tert-Butanol,



CAS No.

690-83-5

TSCA Status

listed on inventory EINECS/ELINCS No.

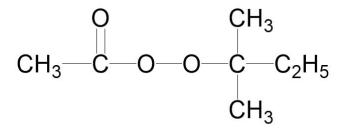
211-715-7

Molecular weight

146.2

Active oxygen content peroxide

6.48%



Tert-Amyl peroxyacetate

Applications

Perodox 133 can be applied in the solution and non-aqueous dispersion polymerization and methacrylates for the manufacture of coatings. In solution polymerization, Perodox 133 is typically applied in combination with another initiator such as dibenzoyl peroxide (Perodox L), tert-Butyl peroxy-2-ethylhexanoate (Perodox 21), or Di-tert-butyl peroxide (Perodox B).

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t^{1/2}$) at various temperatures. For Perodox 133 in chlorobenzene:

0.1 hr at 134°C (273.2°F) 1 hr at 114°C (237.2°F) 10 hr at 96°C (204.8°F)

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $60^{\circ}\text{C} (140^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30° C (86° F)

Note When stored under these recommended storage conditions,

Perodox 133 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 133 is classified as Organic peroxide type D, liquid, Division 5. 2; UN 3105

Major decomposition products

Tert-butanol, acetic acid, di-tert-butyl ether, methane, ethane, etc.



CAS No.

686-31-7

TSCA Status

listed on inventory EINECS/ELINCS No.

211-687-3

Molecular weight

230.3

Active oxygen content peroxide

6.6%-6.95%

tert-Amyl peroxy-2-ethylhexanoate

Applications

Perodox 121 can be used to initiate the (co)polymerization of acrylates and methacrylates in the temperature range between 70°C and 120°C. Perodox 121 can be applied in suspension as well as in solvent and bulk polymerization. Perodox 121 can also be used for the copolymerization of styrene and acrylonitrile and for vinyl acetate based copolymers.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 121 in chlorobenzene:

0.1 hr at 126°C (258.8 °F) 1 hr at 92°C (197.6 °F) 10 hr at 75°C (167 °F)

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 35° C Emergency temperature (T_e) 25° C Control temperature (T_e) 20° C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 5° C

Note When stored under these recommended storage conditions,

Perodox 121 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 121 is classified as Organic peroxide type D; liquid, temperature controlled; Division 5. 2; UN 3115; PG II

Major decomposition products

Carbon dioxide, Methane, tert-Amyl alcohol, Heptane, Ethane, Acetone,



Perodox DA

CAS No.

10508-09-5 TSCA Status

listed on inventory

EINECS/ELINCS No.

234-042-8

Molecular weight

174.3

Active oxygen content peroxide

8.81%

$$CH_3$$
 CH_3
 CH_3 CH_2 CH_3
 CH_3 CH_3

Di-tert-amyl peroxide

Applications

Perodox DA is used as an initiator (radical source)in the polymerisation of monomers, like e.g. ethylene, styrene, and(meth)acrylates.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox DA in chlorobenzene:

0.1 hr at 184°C (363.2 °F) 1 hr at 143°C (289.4 °F) 10 hr at 123°C (253.4 °F)

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 80° C Emergency temperature (T_e) 70° C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30° C

Note When stored under these recommended storage conditions,

Perodox DA will remain within the Do Sender Chem

specifications for a period of at least 6 months after delivery.

Packaging and transport

20 kg.

Perodox DA is classified as Organic Peroxide Type E,liquid,Division 5. 2; UN 3107; PG II

Major decomposition products

Carbon dioxide, Methane, tert-Amyl alcohol, Heptane, Ethane, Acetone,



CAS No.

3006-86-8 TSCA Status

listed on inventory EINECS/ELINCS No.

221-111-2

Molecular weight

260.4

Active oxygen content peroxide

10.58%

1,1-Di(tert-butylperoxy) cyclohexane

Applications

Perodox 22 is an initiator for (co)polymerization of ethylene, styrene and acrylonitrile.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 22 in chlorobenzene:

| 0.1 hr | at 134°C (273°F) |
|-----------|----------------------------|
| 1 hr | at 113°C (235°F) |
| 10 hr | at 94°C (201°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 142.40 kJ/mole |
| A | 3.47E+15 s-1 |
| R | 8.3142 J/mole∙K |
| T | $(273.15 + {}^{\circ}C) K$ |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 60°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 25° C

Note When stored under these recommended storage conditions,

Perodox 22 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 22 is classified as Organic peroxide type C; liquid, Division 5. 2; UN 3103.

Major decomposition products

Carbon dioxide, tert-Butanol, Isomers of isooctane



CAS No.

6731-36-8 TSCA Status

listed on inventory EINECS/ELINCS No.

229-782-3

Molecular weight

302.5

Active oxygen content peroxide

10.58%

Applications

Perodox 29 is an initiator (90% active ingredient in isododecane) for (co)polymerization of ethylene, styrene, acrylonitrile, (meth)acrylates and diethylene glycol bis(allyl carbonate)-based optical monomers.

CH₃

1,1-Di(tert-butylperoxy)-3,3,5-trimethylcyclohexane

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo selfaccelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 60°C

The Heat Accumulation Storage Test is a Method recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 25°C

When stored under these recommended storage conditions, Note

Perodox 29 will remain within the Do Sender Chem

specifications for a period of at least 6 months after delivery.

Packaging and transport

20 kg.

Perodox 29 is classified as Organic peroxide type C; liquid, Division 5. 2; UN 3103.

Major decomposition products

tert-Butanol, poly(3,3,5-trimethylcaprolactone), Acetone, Methane, Water, 3,3,5-Trimethylcyclohexanone



CAS No.

15667-10-4 TSCA Status

listed on inventory EINECS/ELINCS No.

239-741-1

Molecular weight

288.4

Active oxygen

8.76%-8.99%

$$C_{2}H_{5}$$
 $C_{1}H_{5}$
 $C_{2}H_{5}$
 $C_{1}H_{3}$
 $C_{2}H_{5}$
 $C_{1}H_{3}$
 $C_{2}H_{5}$
 $C_{1}H_{3}$
 $C_{2}H_{3}$

1,1-Di(tert-amylperoxy) cyclohexane

Applications

Perodox 122 is a highly efficient midrange initiator for curing unsaturated polyester and vinyl ester resins at elevated temperatures. Typical applications include prepreg, premix, BMC, TMC, SMC and pultrusion. Perodox 122 provides excellent catalyzed shelf-life in BMC and SMC formulations.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 55°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30° C

Note When stored under these recommended storage conditions,

Perodox 122 will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 129 is classified as Organic peroxide type C; liquid, Division 5. 2; UN 3103.

Major decomposition products

Carbon dioxide, cyclohexanone, hexanoic acid, methyl hexanoate,



Perodox TAHP

CAS No.

3425-61-4 TSCA Status

listed on inventory EINECS/ELINCS No.

222-321-7

Molecular weight

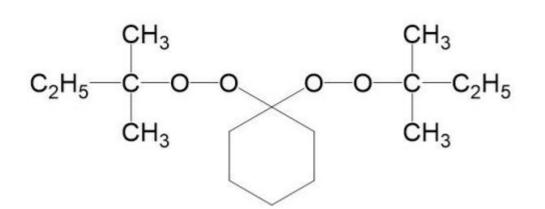
104.1

Active oxygen content peroxide

8.76%-8.99%

Concentration

12.30-13.06%



tert-Amyl hydroperoxide, 85% solution in water

Applications

Perodox TAHP can be used as initiator for the solution (co)polymerization of styrene, acrylates and methacrylates.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox TAHP in chlorobenzene:

0.1 hr at 207°C 1 hr at 185°C 10 hr at 164°C Formula 1 $kd = A \cdot e - Ea/RT$ $t^{1/2} = (\ln 2)/kd$ Formula 2 186.01 kJ/mole Ea 3.18E+17 s-1 Α 8.3142 J/mole·K $(273.15 + {}^{\circ}C) K$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 80°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30° C Ts Min. -5° C

Note When stored under these recommended storage conditions,

Perodox TAHP will remain within the Do Sender Chem

specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox TAHP is classified as Organic peroxide type F; liquid, Division 5. 2; UN 3109.

Major decomposition products

Methane, tert-Amyl alcohol, Ethane, Propane



Perodox TBHP

CAS No.

75-91-2

TSCA Status

listed on inventory EINECS/ELINCS No.

200-915-7

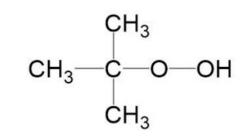
Molecular weight

oo a

90.1

Active oxygen content peroxide

12.71%



Tert-Butyl hydroperoxide

Applications

Perodox TBHP can be used as an initiator in bulk, aqueous solution and emulsion polymerization of styrene, acrylates and methacrylates. The polymerization can be initiated by radicals generated by the thermal decomposition of TBHP70 above 110°C or through a redox mechanism at low temperatures. Effective organic reducing agents are ascorbic acid and sodium formaldehyde sulfoxylate, possibly combined with heavy metal compounds such as cobalt or iron salts. Typical applications of TBHP70 include: manufacture of acrylate, vinylacetate, styrene-butadiene and other latices, curing of styrene polyester resins and use as an oxidizing agent for hydrocarbons or other chemicals.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox TBHP in chlorobenzene:

0.1 hr at 228°C 1 hr at 190°C 10 hr at 153°C

Formula 1 $kd = A \cdot e - Ea/RT$ Formula 2 $t^{1/2} = (\ln 2)/kd$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 80°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 30° C

Ts Min. 0°C to prevent freezing

Note When stored according to these recommended storage

conditions, Perodox TBHP will remain within the Do Sender Chem specifications for a period of at least 3 months after

delivery

Packaging and transport

20 kg.

Perodox TBHP is classified as Organic peroxide type F, liquid; Division 5.2; UN 3109

Major decomposition products

Methane, Acetone, tert-Butanol



CAS No.

70833-40-8 TSCA Status

listed on inventory

EINECS/ELINCS No.

274-919-2

Molecular weight

260.4
Active oxygen content peroxide

6.14%

$$CH_3$$
 $-(CH_2)_3$ $-CH$ $-CH_2$ $-O$ $-C$ $-O$ $-O$ $-C$ $-C_2H_5$ $-CH_3$

tert-Amylperoxy 2-ethylhexyl carbonate

Applications

Perodox 131 can be used to crosslink elastomers at intermediate temperatures, a typical crosslink temperature is 150°C. Perodox 131 is recommended for highly demanding applications e.g. as crosslinker for the polyolefin elastomer copolymer (POE) film in photovoltaic devices.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 131 in chlorobenzene:

0.1 hr at 134°C (273°F) at 113°C (235°F) 1 hr at 95°C (203°F) 10 hr $kd = A \cdot e - Ea/RT$ Formula 1 $t^{1/2} = (\ln 2)/kd$ Formula 2 148.4 kJ/mole Ea 2.22E+16 s-1 Α 8.3142 J/mole·K $(273.15 + {}^{\circ}C) K$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $55^{\circ}C (131^{\circ}F)$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 20° C (68°F)

Note When stored according to these recommended storage

conditions, Perodox 131 will remain within the Do Sender Chem specifications for a period of at least 3 months after

delivery

Packaging and transport

20 kg.

Perodox 131 is classified as Organic peroxide type D, liquid; Division 5. 2; UN 3105.

Major decomposition products

Carbon dioxide, 2-Ethylhexanol, tert-Amyl alcohol



Perodox HMa

CAS No.
37206-20-5
TSCA Status
listed on inventory
EINECS/ELINCS No.
253-396-4

Methyl isobutyl ketone peroxide

Applications

Perodox HMa is a methyl isobutyl ketone peroxide (MIBKP) formulation for curing unsaturated polyester resins as such or in the presence of a cobalt accelerator in the temperature range of 60-150°C.

Perodox HMa is a methyl isobutyl ketone peroxide (MIBKP) formulation for the curing of unsaturated polyester resins as such or in the presence of a cobalt accelerator in the temperature range of 60-150°C. Perodox HMa can be used in combination with a cobalt accelerator (e.g. Accelerator NL-49PN) for those applications where a long gel time or production time is required at ambient temperature and a fast cure at higher temperatures e.g. 60-100°C. Application area can be: filament winding and production of flat and corrugated sheets. Perodox HMa shows also without the addition of a cobalt accelerator a high reactivity at elevated temperatures, which is demonstrated by its low activation temperature. Perodox HMa can therefore successfully be used in the temperature range of approximately 80-150°C: for the production of flat and corrugated sheets where a discoloration by the cobalt accelerator is not acceptable, as kicker for the pultrusion application in combination with a low reactive peroxide like Perodox 29 or Perodox C. The low reactive peroxide must be included in the formulation in order to achieve a complete cure.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 55°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 25°C

Note When stored according to these recommended storage

conditions, Perodox HMa will remain within the Do Sender Chem specifications for a period of at least 3 months after

delivery

Packaging and transport

20 kg.

Perodox HMa is classified as Organic peroxide type D; liquid, Division 5.2; UN 3105.

Major decomposition products

Carbon monoxide, water, mixture of aliphatic acids and ketones



CAS No.

24748-23-0 TSCA Status

listed on inventory EINECS/ELINCS No.

429-320-2

Molecular weight

264.3

3,6,9-Triethyl-3,6,9-trimethyl-1,4,7-triperoxonane

Applications

Perodox 301 is an efficient peroxide formulation for the production of controlled rheology polypropylene (CR-PP) in an extrusion process. It also can be used for the market segments: polymer production, acrylics production and polymer recycling with their different applications/functions.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 301 in chlorobenzene:

| 0.1 hr | 170°C (338°F) |
|-----------|--------------------------|
| 1 hr | 146°C (295°F) |
| 10 hr | 125°C (257°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 150.23 kJ/mole |
| A | 1.02E+15 s-1 |
| R | 8.3142 J/mole·K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $110^{\circ} \text{C} (230^{\circ} \text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. $40^{\circ}\text{C} (104^{\circ}\text{F})$ Ts min. $10^{\circ}\text{C} (50^{\circ}\text{F})$

Note When stored according to these recommended storage conditions, Perodox 301 will remain within the Do Sender Chem specifications for a period of at least 12 months after delivery

Packaging and transport

20 kg.

Perodox 301 is classified as Organic peroxide type D; liquid, Division 5.2; UN 3105.

Major decomposition products

Carbon dioxide, Ethane, Methane, Ethyl acetate, Methyl ethyl ketone, Methyl acetate



Perodox TBMA

CAS No.

1931-62-0

TSCA Status

listed on inventory

EINECS/ELINCS No.

217-691-1

Molecular weight

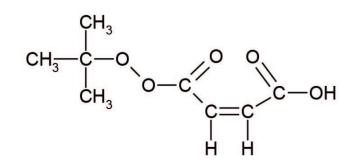
188.2

Active oxygen content peroxide

8.5%

Concentration

3.23-3.57%



tert-Butyl monoperoxymaleate, 40%

Applications

Perodox TBMA is a 40% suspension which is used for room temperature curing of acrylic solid surface without yellow discoloration. It is used in combination with Accelerator CTAS for room temperature cure.

Perodox TBMA can be used as for the ambient temperature curing of MMA resins. Without the addition of an accelerator, the cure of the acrylate resin starts only above the activation temperature of 90°C. For the cure at ambient temperature, Perodox TBMA must be accelerated with a combination of special accelerators. Together with the Accelerators R-913 and Ca(OH)2, a reasonable geltime is obtained with a fast cure at ambient temperatures. The most important advantage of a cure system based on Perodox TBMA is the absence of yellow discoloration and a lower residual monomer content compared to results obtained with a BPO/amine curing system.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $50^{\circ}\text{C} (122^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. $25^{\circ}C (77^{\circ}F)$

Note When stored according to these recommended storage conditions, Perodox TBMA will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery

Packaging and transport

20 kg.

Perodox TBMA is classified as Organic peroxide type E; solid, Division 5.2; UN 3108; PG I.

Major decomposition products

Carbon dioxide, methane, ethane, isononanoic acid, 2-phenylpropanol-2, acetophenone, cumene, etc.



Perodox EHP

CAS No.

16111-62-9

TSCA Status

listed on inventory

EINECS/ELINCS No.

240-282-4

Molecular weight

Di(2-ethylhexyl) peroxydicarbonate

346.5

Active oxygen content peroxide

4.62%

Applications

Perodox EHP is an initiator for the (co)polymerization of ethylene, vinyl chloride, vinylidene chloride, acrylates and methacrylates. Polymerization of ethylene: Perodox EHP is an efficient initiator for the production of Low Density Polyethylene (LDPE). It is used both for tubular and autoclave processes. In most cases a combination with other peroxides is used to ensure a broad reactivity range. Polymerization of vinyl chloride: Perodox EHP is applied as an initiator for the suspension and mass polymerization of vinyl chloride in the temperature range between 40°C and 65°C. Perodox EHP can be used alone or in combination with other peroxides, such as 1,1,3,3-Tetramethylbutyl peroxyneodecanoate Perodox 423), Cumyl peroxyneodecanoate (Perodox 99) or Dilauroyl peroxide (LUNA), to increase reactor efficiency.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox EHP in chlorobenzene:

| 0.1 hr | at 83°C (181°F) |
|-----------|--------------------------|
| 1 hr | at 64°C (147°F) |
| 10 hr | at 47°C (117°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 122.45 kJ/mole |
| A | 1.83E+15 s-1 |
| R | 8.3142 J/mole·K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $0^{\circ}\text{C }(32^{\circ}\text{F})$ Emergency temperature (T_e) $-10^{\circ}\text{C }(14^{\circ}\text{F})$ Control temperature (Tc) $-20^{\circ}\text{C }(-4^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. $-20^{\circ}\text{C} (-4^{\circ}\text{F})$

Note When stored according to these recommended storage conditions, Perodox EHP will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery

Packaging and transport

20 kg.

Perodox EHP is classified as Organic peroxide type C; liquid, temperature controlled; Division 5.2; UN 3113.

Major decomposition products

Carbon dioxide, 2-Ethylhexanol



CAS No.

15520-11-3 TSCA Status

listed on inventory

EINECS/ELINCS No.

239-557-1

398.5

Molecular weight

Di(4-tert-butylcyclohexyl) peroxydicarbonate

Applications

Perodox 16 is applied as an initiator for the suspension and mass polymerization of vinyl chloride in the temperature range between 40°C and 65°C.Perodox 16 can be used alone or in combination with other peroxides, such as 1,1,3,3-Tetramethylbutyl peroxyneodecanoate, Cumyl peroxyneodecanoate) or Dilauroyl peroxide (Perodox LUNA), to increase reactor efficiency

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t1/2) at various temperatures. For Perodox 16 in chlorobenzene:

0.1 hr 82°C (180°F) 64°C (147°F) 1 hr 48°C (118°F) 10 hr Formula 1 $kd = A \cdot e - Ea/RT$ $t^{1/2} = (\ln 2)/kd$ Formula 2 126.39 kJ/mole Ea 7.44E+15 s-1 Α 8.3142 J/mole·K $(273.15 + {}^{\circ}C) K$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo selfaccelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

40°C **SADT** Emergency temperature (T_e) 35°C Control temperature (Tc) 30°C

The Heat Accumulation Storage Test is a Method recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

20°C (please see note below) Ts max.

When stored under the recommended storage conditions, Perodox Note 16 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery. The Ts max of 20°C is not to be interpretated as ambient or room temperature as this differs per region and season. Perodox 16 has a high quality composition and to hold that it should be stored at below 20°C. At temperatures above 20°C the decomposition of Perodox 16 progresses fast which leads to significant loss of quality.

Packaging and transport

20 kg.

Perodox 16 is classified as Organic peroxide type C; solid, temperature controlled; Division 5. 2; UN 3114.

Major decomposition products

Carbon dioxide, 4-tert-Butyl-cyclohexanol



CAS No.

3851-87-4 TSCA Status

listed on inventory

EINECS/ELINCS No.

223-356-0

Molecular weight

314.5

Active oxygen content peroxide

5.09%

Di(3,5,5-trimethylhexanoyl) peroxide, 75%

Applications

Perodox 36 is an efficient initiator (75% active ingredient in odorless mineral spirits) for (co)polymerization of ethylene, vinyl chloride, vinylidene chloride acrylates and methacrylates. Polymerization of ethylene: Perodox 36 is an efficient initiator for the ethylene polymerization under high pressure in both autoclave and tubular processes. To obtain a wide spectrum of polymerization temperatures, combinations with other peroxides are applied in practice. Polymerization of vinyl chloride: Perodox 36 can also be applied as an initiator for the suspension polymerization of vinyl chloride in the temperature range 50-70°C.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 36 in chlorobenzene:

0.1 hr at 96°C (205°F) at 77°C (171°F) 1 hr at 59°C (138°F) 10 hr $kd = A \cdot e - Ea/RT$ Formula 1 Formula 2 $t\frac{1}{2} = (\ln 2)/kd$ 128.34 kJ/mole Ea 2.84E+15 s-1 8.3142 J/mole·K (273.15+°C) K Т

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $20^{\circ}\text{C }(68^{\circ}\text{F})$ Emergency temperature (T_e) $10^{\circ}\text{C }(50^{\circ}\text{F})$ Control temperature (Tc) $0^{\circ}\text{C }(32^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. 0°C

Ts min. -25°C to prevent crystallization

Note When stored according to these recommended storage conditions, Perodox 36 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 36 is classified as Organic peroxide type D; liquid, temperature controlled, Division 5. 2; UN 3115.

Major decomposition products

Carbon dioxide, 2,2,4,7,9,9-Hexamethyldecane, 2,4,4-Trimethylpentane



CAS No.

29240-17-3

TSCA Status

listed on inventory

EINECS/ELINCS No.

64741-65-7

Molecular weight

188.3 Active oxygen content peroxide

8.5%

$$CH_{3}O$$
 CH_{3} CH_{3} CH_{3} CH_{3} CH_{3} CH_{3} CH_{3} CH_{3}

tert-Amyl peroxypivalate, 75%

Applications

Perodox 75 is an initiator (75% active ingredient in isododecane) for (co)polymerization of ethylene, styrene, vinyl chloride, vinylidine chloride and polyols. Polymerization of ethylene: Perodox 75 is an efficient initiator for the ethylene polymerization under high pressure in both autoclave and tubular processes. To obtain a wide spectrum of polymerization temperatures, combinations with other peroxides are applied in practice. Polymerization of vinyl chloride: Perodox 75 can be applied as an initiator forthe suspension polymerization of vinyl chloride in the temperature range between 50°C and 65°C Perodox 125 may be used in combination with other initiators to increase reactor efficiency.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 125 in chlorobenzene:

| 0.1 hr | at 91°C (196°F) |
|-----------|--------------------------|
| 1 hr | at 72°C (162°F) |
| 10 hr | at 54°C (129°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 127.76 kJ/mole |
| A | 4.12E+15 s-1 |
| R | 8.3142 J/mole∙K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $25^{\circ}\text{C} (77^{\circ}\text{F})$ Emergency temperature (T_e) $15^{\circ}\text{C} (59^{\circ}\text{F})$ Control temperature (Tc) $10^{\circ}\text{C} (50^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. -10° C (14° F)

Ts Min. -30° C (-22°F) to prevent crystallization

Note When stored according to these recommended storage conditions, Perodox 125 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 125 is classified as Organic peroxide type C; liquid, temperature controlled, Division 5.2; UN 3113.

Major decomposition products

Carbon dioxide, Ethane, Isobutane, Isobutene, Acetone, Methyl ethyl ketone, tert-Amyl alcohol



CAS No.

26748-41-4

TSCA Status

listed on inventory

EINECS/ELINCS No.

247-955-1

Molecular weight

244.4 Active oxygen content peroxide

6.55%

$$R_{1} O CH_{3}$$
 $R_{2} CH_{3}$
 CH_{3}
 CH_{3}

tert-Butyl peroxyneodecanoate, 75%

Applications

Perodox 23 is an efficient initiator for the production of Low Density Polyethylene (LDPE). It is used both for tubular and autoclave processes. In most cases a combination with other peroxides is used to ensure a broad reactivity range. Perodox 23 is used as an initiator for the suspension polymerization of vinyl chloride in the temperature range between 40°C and 65°C. Perodox 23 can be used alone or in combination with other peroxides, such as 1,1,3,3-Tetramethylbutyl peroxyneodecanoate (Perodox 423), Cumyl peroxyneodecanoate (Perodox 99) or Dilauroyl peroxide (LUNA), to increase reactor efficiency.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 23 in chlorobenzene:

| 0.1 hr | at 84°C (183°F) |
|-----------|--------------------------|
| 1 hr | at 64°C (147°F) |
| 10 hr | at 46°C (115°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 115.47 kJ/mole |
| A | 1.52E+14 s-1 |
| R | 8.3142 J/mole·K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $20^{\circ}\text{C }(68^{\circ}\text{F})$ Emergency temperature (T_e) $10^{\circ}\text{C }(50^{\circ}\text{F})$ Control temperature (Tc) $0^{\circ}\text{C }(32^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. -10° C (14°F)

Ts min. -20°C (-4°F) to prevent crystallization

Note When stored according to these recommended storage conditions, Perodox 23 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 23 is classified as Organic peroxide type D; liquid, temperature controlled, Division 5. 2; UN 3115.

Major decomposition products

Carbon dioxide, tert-Butanol, Isomers of neononane



CAS No.

927-07-1

TSCA Status

listed on inventory

EINECS/ELINCS No.

213-147-2

Molecular weight

174.2 Active oxygen content peroxide

9.18%

Concentration

6.89%

$$CH_{3}$$
 CH_{3} CH_{3} CH_{3} CH_{3} CH_{3} CH_{3} CH_{3}

tert-Butyl peroxypivalate, 75%

Applications

Perodox 25 is an initiator (75% active ingredient in isoparrafin) for (co)polymerization of acrylates and methacrylates. It can be used for the market segments: polymer production and acrylics with their different applications/functions

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 25 in chlorobenzene:

0.1 hr at 94°C (201°F) at 75°C (167°F) 1 hr at 57°C (135°F) 10 hr Formula 1 $kd = A \cdot e - Ea/RT$ $t^{1/2} = (\ln 2)/kd$ Formula 2 123.59 kJ/mole Ea 7.09E+14 s-1 Α 8.3142 J/mole·K $(273.15 + {}^{\circ}C) K$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $20^{\circ}\text{C }(68^{\circ}\text{F})$ Emergency temperature (T_e) $10^{\circ}\text{C }(50^{\circ}\text{F})$ Control temperature (Tc) $0^{\circ}\text{C }(32^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. $-5^{\circ}C (23^{\circ}F)$

Ts min. - 15° C (5° F) to prevent crystallization

Note When stored according to these recommended storage conditions, Perodox 25 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 25 is classified as Organic peroxide type C; liquid, temperature controlled, Division 5. 2; UN 3113.

Major decomposition products

Carbon dioxide, Isobutene, tert-Butanol, Acetone, Isobutane



Perodox 44B

CAS No.

37187-22-7 (for EU: 13784-51-5)

TSCA Status

listed on inventory EINECS/ELINCS No.

253-384-9 (for EU: 237-438-9)

Acetyl acetone peroxide

Applications

Perodox 44B is an acetylacetone peroxide formulation for fast curing of unsaturated polyester resins in the presence of a cobalt accelerator at room and elevated temperatures.

Perodox 44B is an acetyl acetone peroxide formulation for the curing of unsaturated polyester resins in the presence of a cobalt accelerator at room and elevated temperatures. With the curing system Perodox 44B /cobalt accelerator a much faster speed of cure may be achieved than with curing systems based on a MEKP plus cobalt accelerator, at room and elevated temperatures. Normally the gel times with Perodox 44B are comparable to those with Perodox M-50.

Perodox 44B is particularly suitable in those applications where a fast mold-turnover is required, e.g. for the cold press molding or resin injection molding techniques. The system Perodox 44B /cobalt accelerator will give a higher peak exotherm than a standard MEKP/cobalt accelerator system. Due to this fact, is it recommendable to avoid the production of too thick laminates in one operation. At low temperatures a reasonable speed of cure is still obtained when Perodox 44B is used in combination with large amounts of cobalt accelerator possibly in combination with N,N Dimethylaniline as promotor.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 60°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. 25° C Ts min. - 10° C

Note When stored according to these recommended storage conditions, Perodox 25 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 44B land and sea transport is classified as Organic peroxide type E; liquid, Division 5.2; UN 3107. Air transport is classified as Organic peroxide type D; liquid, Division 5.2; UN 3105.

Major decomposition products

Carbon dioxide, acetyl acetone, mixture of aliphatic acids, water



CAS No.

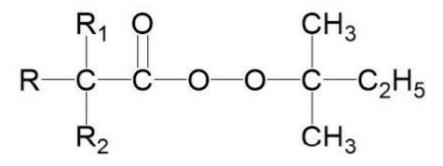
3025-88-5 TSCA Status

listed on inventory EINECS/ELINCS No.

221-184-0

Active oxygen

4.58-4.71 %



tert-Amyl peroxyneodecanoate

Applications

Perodox 123 is an initiator (75% active ingredient in odorless mineral spirits) for (co)polymerization of ethylene, styrene, vinyl chloride, vinylidine chloride, acrylates and methacrylates.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 123 in chlorobenzene:

| 0.1 hr | at 81°C (178°F) |
|-----------|--------------------------|
| 1 hr | at 61°C (142°F) |
| 10 hr | at 43°C (109°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 114.38 kJ/mole |
| A | 1.47E+14 s-1 |
| R | 8.3142 J/mole·K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $20^{\circ}\text{C }(68^{\circ}\text{F})$ Emergency temperature (T_e) $10^{\circ}\text{C }(50^{\circ}\text{F})$ Control temperature (Tc) $0^{\circ}\text{C }(32^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. -15° C (5° F)

Ts Min. - 25°C (-13°F) to prevent crystallization

Note When stored according to these recommended storage conditions, Perodox 123 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 123 is classified as Organic peroxide type D, liquid, temperature controlled, Division 5.2; UN 3115.

Major decomposition products

Carbon dioxide, Ethane, Isobutane, Isobutene, Acetone, Methyl ethyl ketone, tert-Amyl alcohol.



CAS No.

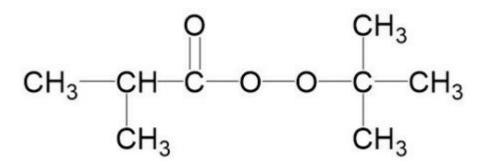
109-13-7 TSCA Status

listed on inventory EINECS/ELINCS No.

203-650-5

Active oxygen

4.89-5.09 %



tert-Butyl peroxyisobutyrate

Applications

Perodox 41 is an initiator (50% active ingredient in odorless mineral spirits) for (co)polymerization of ethylene and styrene. Perodox 41 is an efficient initiator for the ethylene polymerization under high pressure in both autoclave and tubular processes. To obtain a wide spectrum of polymerization temperatures, combinations with other peroxides are applied in practice. Depending on reaction conditions, Perodox 41 is active in the temperature range between 165°C and 250°C.Perodox 41 can be used for the polymerization of styrene in the temperature range between 85°C and 105°C. Main application is the copolymerization of styrene with acrylonitrile, acrylates and methacrylates. Perodox 41 can be used as initiator for the solution (co)polymerization of acrylates and methacrylates in the temperature range of 80-160°C, amongst others for the manufacture of coatings. Perodox 41 can also be applied as an initiator for the bulk and suspension (co)polymerization of acrylates and methacrylates.

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 41 in chlorobenzene:

| 0.1 hr | at 118°C (244°F) |
|-----------|------------------|
| 1 hr | at 98°C (208°F) |
| 10 hr | at 79°C (174°F) |
| Formula 1 | kd=A·e-Ea/RT |
| Formula 2 | t½=(ln2)/kd |
| Ea | 135.16 kJ/mole |
| A | 2.07E+15 s-1 |
| R | 8.3142 J/mole·K |
| T | (273.15+°C) K |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $30^{\circ}\text{C }(86^{\circ}\text{F})$ Emergency temperature (T_e) $20^{\circ}\text{C }(68^{\circ}\text{F})$ Control temperature (Tc) $15^{\circ}\text{C }(59^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 10° C (50° F)

Note When stored according to these recommended storage conditions, Perodox 41 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 41 is classified as Organic peroxide type D; liquid, temperature controlled; Division 5.2; UN 3115.

Major decomposition products

Carbon dioxide, Propane, Propene, tert Butanol, 2-Isopropoxy-2-methylpropane



CAS No.

13052-09-0
TSCA Status CH_3 CH_3

EINECS/ELINCS No.

2,5-Dimethyl-2,5-di(2-ethylhexanoylperoxy) hexane

Assay

≥ 92.0 %

Applications

In suspension polymerization processes, Perodox 141 can be used for the polymerization of styrene at approximately 90°C. Perodox 141 has an activity comparable with Dibenzoyl peroxide (Perodox L75). However, the bifunctional Perodox 141 gives higher molecular weights of the polymer. Applying the same molecular weights means a reduction of polymerization time

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $35^{\circ}\text{C }(95^{\circ}\text{F})$ Emergency temperature (T_{e}) $25^{\circ}\text{C }(77^{\circ}\text{F})$ Control temperature (T_{c}) $20^{\circ}\text{C }(68^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. 15° C

Note When stored according to these recommended storage conditions, Perodox 141 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 141 is classified as Organic peroxide type C; liquid, temperature controlled, Division 5. 2; UN 3113.

Major decomposition products

Carbon dioxide, Acetone, 2-Pentanone, Heptane, Heptenes, tert-Amyl alcohol, 2,5-Bis(1-ethylpentoxy)-2,5-dimethylhexane, 2,5-Dimethyl-2,5-hexanediol



CAS No.

1068-27-5

TSCA Status

listed on inventory EINECS/ELINCS No.

213-944-5

Molecular weight

$$CH_3$$
 CH_3 CH_3

2,5-Dimethyl-2,5-di(tert-butylperoxy) hexyne-3

286.4

Active oxygen content peroxide

11.17%

Concentration

9.27-9.61%

Applications

Perodox 145 is a bifunctional peroxide which is used for the crosslinking of natural rubber and synthetic rubbers, as well as polyolefins. Safe processing temperature: 145°C (rheometer ts2 > 20 min.). Typical crosslinking temperature: 185°C (rheometer t90 about 12 min.).

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT 80°C

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. 30° C (86°F) Ts min. 10° C (50°F)

Note When stored according to these recommended storage conditions, Perodox 145 will remain within the Do Sender Chem specifications for a period of at least 6 months after delivery.

Packaging and transport

20 kg.

Perodox 145 is classified as Organic peroxide type C; liquid, Division 5. 2; UN 3103.

Major decomposition products

Methane, 2,5-Dimethyl-3-hexyn-2,5-diol, Acetone, tert-Butanol



Perodox D

CAS No.

2167-23-9 TSCA Status

listed on inventory EINECS/ELINCS No.

218-507-2

Molecular weight

234.3

Active oxygen content peroxide

13.66%

Concentration

6.69-6.97%

2,2-Di(tert-butylperoxy) butane

Applications

Perodox D is an initiator (50% active ingredient in mineral oil) for (co)polymerization of styrene, acrylonitrile, acrylates and methacrylates. Polymerization of styrene Perodox D can be used for the suspension polymerization and copolymerization of styrene in the temperature range between 90°C and 130°C. During polymerization the temperature is increased in steps. A comparison is made between thermally and Perodox D initiated (0.08 mmole/100 g styrene) polymerization of styrene in a mass process, applying a typical temperature scheme (110-140-180°C).

Half-life data

The reactivity of an organic peroxide is usually given by its half-life (t1/2) at various temperatures. For Perodox D in chlorobenzene:

0.1 hr at 136°C at 116°C 1 hr at 98°C 10 hr

Formula 1 $kd = A \cdot e - Ea/RT$ $t^{1/2} = (\ln 2)/kd$ Formula 2 154.08 kJ/mole Ea 9.30E+16 s-1 Α 8.3142 J/mole·K $(273.15 + {}^{\circ}C) K$

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo selfaccelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

80°C SADT

The Heat Accumulation Storage Test is a Method recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

When stored according to these recommended storage Note conditions, Perodox D will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox D is classified as Organic peroxide type C; liquid, Division 5. 2; UN 3103.

Major decomposition products

Methane, Ethane, Acetone, tert-Butanol, Carbon dioxide, 2-Methoxy-2-methylpropane



CAS No.

53220-22-7

TSCA Status

listed on inventory

EINECS/ELINCS No.

258-436-4

Molecular weight

514.8

Dimyristyl peroxydicarbonate

Applications

Perodox 26 is an initiator for (co)polymerization of vinyl chloride, vinylidene chloride and (meth)acrylates. Perodox 26 is applied as an initiator for the suspension and mass polymerization of vinyl chloride in the temperature range of 40-65°C. Light yellow transparent antifouling agent particularly suited for protection of internal walls of reactors used for suspension polymerization of PVC.

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $35^{\circ}\text{C }(95^{\circ}\text{F})$ Emergency temperature (T_e) $25^{\circ}\text{C }(77^{\circ}\text{F})$ Control temperature (Tc) $20^{\circ}\text{C }(68^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts max. 15° C (59° F)

Note When stored according to these recommended storage conditions, Perodox 26 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 26 is classified as Organic peroxide type D; solid, temperature controlled, Division 5. 2; UN 3116.

Major decomposition products

Carbon dioxide, Tetradecanol



CAS No.

26322-14-5 TSCA Status

listed on inventory

EINECS/ELINCS No.

247-611-0

Molecular weight

570.9

Active oxygen content peroxide

2.80%

Concentration

2.52-2.61%

Applications

Perodox 24 is an initiator for (co)polymerization of vinyl chloride, vinylidene chloride, acrylates and methacrylates. Used to increase melt strength of polypropylene to make high melt strength polypropylene (HMS-PP) by reactive extrusion process.

Dicetyl peroxydicarbonate

Half-life data

The reactivity of an organic peroxide is usually given by its half-life ($t\frac{1}{2}$) at various temperatures. For Perodox 24 in chlorobenzene:

| 0.1 hr | at 84°C (183°F) |
|-----------|----------------------------|
| 1 hr | at 65°C (149°F) |
| 10 hr | at 48°C (118°F) |
| Formula 1 | $kd = A \cdot e - Ea/RT$ |
| Formula 2 | $t^{1/2} = (\ln 2)/kd$ |
| Ea | 124.30 kJ/mole |
| A | 3.02E+15 s-1 |
| R | 8.3142 J/mole∙K |
| T | $(273.15 + {}^{\circ}C) K$ |
| | |

Thermal stability

Organic peroxides are thermally unstable substances, which may undergo self-accelerating decomposition. The lowest temperature at which self-accelerating decomposition of a substance in the original packaging may occur is the Self Accelerating Decomposition Temperature (SADT). The SADT is determined on the basis of the Heat Accumulation Storage Test.

SADT $40^{\circ}\text{C} (104^{\circ}\text{F})$ Emergency temperature (T_e) $35^{\circ}\text{C} (95^{\circ}\text{F})$ Control temperature (Tc) $30^{\circ}\text{C} (86^{\circ}\text{F})$

Method The Heat Accumulation Storage Test is a recognized test method for the determination of the SADT of organic peroxides (see Recommendations on the Transport of Dangerous Goods, Manual of Tests and Criteria - United Nations, New York and Geneva).

Storage

Due to the relatively unstable nature of organic peroxides a loss of quality can be detected over a period of time. To minimize the loss of quality, Do Sender Chem recommends a maximum storage temperature (Ts max.) for each organic peroxide product.

Ts Max. $20^{\circ} \text{C } (68^{\circ} \text{F})$

Note When stored according to these recommended storage conditions, Perodox 24 will remain within the Do Sender Chem specifications for a period of at least 3 months after delivery.

Packaging and transport

20 kg.

Perodox 24L is classified as Organic peroxide type F; solid, temperature controlled, Division 5. 2; UN 3120.

Major decomposition products

Carbon dioxide, Hexadecanol

